



DPF

| Video Solution on Website:- | | https://physicsaholics.com/home/courseDetails/46 | | | | |
|-------------------------------|--|---|--|--|--|--|
| video Solution on YouTube:- | | https://youtu.be/veQ1_ryXN94 | | | | |
| Vritten Solution on Website:- | | https://physicsaholics.com/note/notesDetalis/19 | | | | |
| Q 1. | (a) Temperature of ta(b) Intensity of elect(c) K.E. of electrons | K - rays decreases, when arget is increased ron beam is increased striking the target is increased striking the target is decreased | | | | |
| Q 2. | | length of X-rays produced in an X-ray tube is λ when the V. What is the minimum wavelength of the X-rays when the V/2? | | | | |
| Q 3. | X-rays are being proproduced in vacuum (a) 3×10^8 m/s (c) 3.1×10^8 m/s | oduced in a tube operating at 10^5 V. The velocity of X-rays tube is (b) 2.8×10^8 m/s (d) 3×10^{10} m/s | | | | |
| Q 4. | The X-rays produced wavelength of (a) 3.09×10^{-8} m (c) 4.09×10^{-8} m | d in a Coolidge tube of potential difference 40 V have minimum (b) $5.09 \times 10^8 \text{m}$ (d) $1.09 \times 10^8 \text{m}$ | | | | |
| Q 5. | A metal block is exp wavelength penetrate (a) 2 Å (c) 6 Å | cosed to beams of X-ray of different wavelength. X-rays of which the most (b) 4 Å (d) 8 Å | | | | |
| Q.6 | emitted from the targ | | | | | |
| Q 7. | Which of the follows (a) 10000 Å (c) 1 Å | ing wavelength falls in X - ray region? (b) 1000 Å (d) 10^{-2} Å | | | | |



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- In X-ray tube, when the accelerating voltage V is doubled, the difference between Q 8. the wavelength of K_{α} line and minimum wavelength of continuous X-ray spectrum
 - (a) Remains constant
- (b) becomes half
- (c) Becomes more than two times
- (d) Becomes less than two times
- Q 9. Mosley measured the frequency (f) of the characteristic X-ray from many metals of different atomic number (Z) and represented his results by a relation known as Moseley's law. This law is (a, b are constant)
 - (a) $f = a(Z b)^2$

(c) $f^2 = a(Z - b)$

- (b) $Z = a(f b)^2$ (d) $f = a(Z b)^{1/2}$
- Q 10. If the frequency of K_{α} X-rays emitted from the element with atomic number 31 is v, then the frequency of K_{α} X-rays emitted from the element with atomic number 51 would be
 - (a) $\frac{3}{5}$ v

 $(c)^{\frac{5}{25}}v$

- (b) $\frac{51}{31}$ v (d) $\frac{9}{25}$ v
- Q 11. The X-ray wavelength of L_{α} line of Platinum (Z=78) is 1.30 Å. The X-ray wavelength of L_{α} line of Molybdenum (Z = 42) is (constant b = 7.4)
 - (a) 5.41 Å

(b) 4.20 Å

(c) 2.70 Å

- (d) 1.35 Å
- Q 12. An X-ray tube operates at 40 kV. Suppose the electron converts 70% of its energy into a photon at each collision. Find the 2nd lowest wavelengths emitted from the tube. Neglect the energy imparted to the atom with which the electron collides
 - (a) 44.28 pm

(b) 147.61 pm

(c) 493 pm

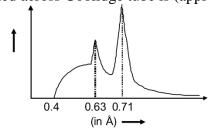
- (d) 122.43 pm
- Q 13. The wavelength of ka X-rays produced by an X rays tube is 0.76Å. The atomic number of the anode material of the tube is
 - (Considering that Bohr's model is applicable)
 - (a) 57

(b) 39

(c) 34

- (d) 41
- Q 14. Figure shows intensity versus wavelength graph of X-rays coming from Coolidgetube with molybdenum as target element. The two peaks shown in graph correspond to K_{α} and K_{β} X-rays.

Wavelength of L_{α} X- rays from Coolidge tube will be (approximately) and Voltage applied across Coolidge tube is (approximately)





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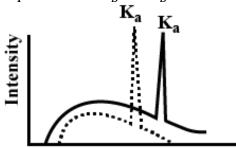
(a) 1.2 Å, 20 kV

(b) 2100 Å, 16 kV

(c) 5.6 Å, 31 kV

(d) 12.3 Å, 18 kV

Q 15. Given curve shows the intensity wavelength relation of X-rays coming from two different Coolidge tubes A and B. The dark curve represents the relation for the tube A in which potential difference between the target and the filament is V_A and the atomic number of the target material is Z_A . Similarly dotted curve is for tube B. Respective quantities are V_B and Z_B for the tube B. Then



Wavelength

(a)
$$V_A > V_B, Z_A > Z_B$$

(b)
$$V_A > V_B$$
, $Z_A < Z_B$

(b)
$$V_A > V_B$$
, $Z_A < Z_B$
(c) $V_A < V_B$, $Z_A > Z_B$
(d) $V_A < V_B$, $Z_A < Z_B$

$$(d) V_A < V_B, Z_A < Z_B$$

Answer Key

| Q.1 c | Q.2 c | Q.3 a | Q.4 a | Q.5 a |
|--------|--------|--------|--------|--------|
| Q.6 a | Q.7 c | Q.8 d | Q.9 a | Q.10 c |
| Q.11 a | Q.12 b | Q.13 d | Q.14 c | Q.15 b |